

Novel corrosion protection method by microbial extracellular polymeric substances (EPS)

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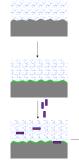
Objectives

Corrosion of metal substrates in aqueous media often is demonstrably induced or sped up by bacterial biofilm /1,2/. *Microbially influenced corrosion (MIC)* is discussed to contribute about 20% of the entire costs of corrosion failures /3/. Among harmful microorganisms one finds so called sulphate reducing bacteria (SRB) - a group of anaerobically existing bacteria producing hydrogen sulphide.

On the other hand, some microorganisms can release chemical substances that were found to influence the adhesive properties of microbial cells - extracellular polymeric substances (EPS) in general highly heterogeneous mixtures of different substance classes /4/. Appropriate EPS are promising candidates for the inhibition of cell adhesion thus for the prevention of biofilm formation. Also biofilms can exhibit protective properties /5/.

The aims of this project are to identify appropriate EPS and to establish a system for corrosion protection by application of EPS - biogenic and renewable substances as protecting films.

Fundamentals



Any surface exposed to a liquid or gaseous phase will quickly be covered by substances present in the phase. The result is called conditioning film and can influence further adhesion of microorganisms.

Biofilms and Biocorrosion

Microbial cells - wrapped in EPS - can adhere on the surface and establish stable bondings depending on the forces of interaction: The first step of biofilm-formation. EPS are constantly released into the evolving biofilm. Whereas EPS are discussed to mediate these forces, no doubts exist on the high importance of EPS for the stability of the biofilm.

The biofilm grows due to adhesion of other species and cell division. Inside the biofilm one also finds metabolites, corrosion products etc





Desorption of cells is induced by addition of

small amounts of special EPS of surface active properties Footprint effect /7/: Upon desorption some bacteria leave patches of substances on the surface.



- resistance against microbial degradation non-corrosive behaviour sufficient adhesiveness
- The high number of synergistically existing microbial cells and the diversity of metabolites can attack the substrate and severely damage the entire system: biocorrosion and MIC. A typical failure mode of microbial attack is local corrosion

	Experimental and Results	
Substrate metals: • pure iron (ARMCO) • carbon steel (St 37) • high alloyed steel (DIN 1.4301) Abrasion with emery paper (P180), rinsing in distilled water, ultrasonication in ethanol, rinsing in acetone,	Setup for mimicking MIC: Desulfovibrio vulgaris as representative sulphate reducing bacteria (SRB) in Postgate C growth medium: mainly sulphate and lactate. • in-situ Inoculation after mounting of the substrates • Pre-inoculated solutions: approx. 5*10 ³ cells*ml ⁻¹ . EPS were harvested from planktonic biofilms by centrifugation and purified. Freeze-dried EPS are	Test of resista SRB-growth medium Postgate was prepared replacing lactate a source by EPS. The planktonic ce of <i>D. Vulgaris</i> – me
Layer preparation on 1.4301 Dip-coating for 2 hours in EPS suspension (5 mg*ml ⁻¹), sterile and anaerobic conditions. Afterwards drying in air or nitrogen atmosphere	re-suspended. Adhesion of <i>D.Vulgaris</i> on 1.4301 modified with <i>D.Vulgaris</i> -EPS Dip-coated surface, in-situ inoculation 1 day 3 days 7 days Fluorescence microscopic images after Lektin staining Uncovered regions (dark spots)	after 4 days of in – are used as ind Result: Only xanthan ai EPS of <i>Lactoba</i> <i>Fermentum</i> can used as carbon
Droplet-coating: 50 µl of EPS-containing suspension are applied on the substrate surface, sterile and anaerobic conditions. Liquid evaporates in air or nitrogen atmosphere A, B: Fluorescence microscopic images after Lektin staining C, D: Atomic force microscopic images Dip-coating: often not evenly shaped laver, dissolves within hours.	even after 7 days! Droplet-coated surface, pre-inoculated, after 5h Droplet-coated surface, pr	Only few EPS can Corrosion due to EPS for <i>D. Vulgaris</i> -EP Up to now poorly adh dissolve in aqueous In Lower cell density <i>D. Vulgaris</i> -EPS Positive effect o xanthan and <i>D. V</i>
ayer, dissolves within hours. Droplet: thicker and more uniform layer exhibiting trenches, more stable against desorption	coated; C, F: uncoated region Single cells clearly distinguishable. Low cell number in regions covered by EPS!	Cher Microscopio Amelio

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ance against microbial degradation

Other cells avoid to adhere

on these patches.

еC as carbon

ell densities neasured cubation dicator.

nd cillus be source.

available I numbers too

Postgate C + Xantha 3,8x10^s ate C + EPS P. flu tgate C + EPS P. f nd No lactate, EPS, stgate C + X 6,2x10^s Postgate C + EPS P. ck ate C + EPS P. fr 2,6x10 2.9x10

Cells*ml⁻¹

Conclusions

h be metabolized by Desulfovibrio vulgaris as carbon source

S under anaerobic conditions only was observed PS and D.Indonesiensis-EPS (AiF 173 ZN).

Ihesive properties! Most EPS quickly desorb or s media within a couple of days or even hours.

nfluence on cell adhesion:

ty and irregularly formed biofilm in case of S when compared to untreated surface. of binary EPS by sequential deposition of Vulgaris-EPS has to be further examined.

Prospectives

emical analysis of EPS-composition ic analysis of interactions EPS | substrate lioration of adhesive properties of EPS Electrochemical investigations on the influence of EPS on (bio-)corrosion

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